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Publisher Taylor & Francis

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## Critical Reviews in Analytical Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713400837>

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**To cite this Article** Hanrahan, Grady and Lu, Kenneth(2006) 'Application of Factorial and Response Surface Methodology in Modern Experimental Design and Optimization', Critical Reviews in Analytical Chemistry, 36: 3, 141 – 151

**To link to this Article:** DOI: 10.1080/10408340600969478

**URL:** <http://dx.doi.org/10.1080/10408340600969478>

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# Application of Factorial and Response Surface Methodology in Modern Experimental Design and Optimization

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**This article critically examines the use of factorial and response surface methodology in modern experimental design and optimization. A survey of important screening and optimization techniques in the literature since 2000 are presented. Current applications in biological, environmental and pharmaceutical analysis, food technology and industrial-related processes are examined.**

**Keywords** experimental design, factorial designs, optimization, response surface methodology, screening

## INTRODUCTION

The importance of, and theoretical concepts behind, experimental design and optimization methodology in research and development efforts has been thoroughly discussed in a number of informative publications (1–17). The two main applications of experimental design are screening, in which the factors that influence the experiment are identified, and optimization, in which the optimal settings or conditions for an experiment are found. The usual approach is to start with a screening design including all controllable factors that may possibly influence the experiment, identify the most important ones, and proceed with an experimental optimization design. Models generated can be evaluated by the analysis of variance (ANOVA). The one-way ANOVA allows experimenters to compare several groups of observations. A two-way ANOVA allows one to study the effects of two factors separately as well as their interactive effects.

Although the choice of an experimental design ultimately depends on the objectives of the experiment and the number of factors to be investigated, initial experimental planning (as shown in Figure 1) is paramount. Screening techniques such as factorial designs allow the experimenter to select which factors are significant and at what levels. The most general (two-level design) is a full factorial design and described as  $2^k$ -designs where the base 2 stands for the number of factor levels and  $k$  the number of factors each with a high and low value (3, 8, 11, 17). The lower level is usually indicated with a ‘–’ sign; the higher

level with a ‘+’ sign. With two factors, this defines a square in the factor space, and with three factors this defines a cube (17). Fractional factorial designs are good alternatives to a full factorial design, especially in the initial stage of a project, and considered a carefully prescribed and representative subset of a full factorial design (11, 17). In fractional factorial designs, the number of experiments is reduced by a number  $p$  according to a  $2^{k-p}$  design.

Response surface methodologies are multivariate techniques that mathematically fit the experimental domain studied in the theoretical design through a response function (1, 2). The two most common designs generally used in response surface modeling are central composite and Box–Behnken designs. In these designs the inputs take on three or five distinct levels, but not all combinations of these values appear in the design. Central composite designs contain imbedded factorial or fractional factorial designs with center points that are augmented with a group of axial (star) points that allow estimation of curvature (17). A central composite design always contains twice as many star points as there are factors in the design (11). The star points represent new extreme values (low and high) for each factor in the design.

The Box–Behnken design is considered an efficient option in response surface methodology and an ideal alternative to central composite designs (11, 17). It has three levels per factor, but avoids the corners of the space, and fills in the combinations of center and extreme levels. Overall, it combines a fractional factorial with incomplete block designs in such a way as to avoid the extreme vertices and to present an approximately rotatable design with only three levels per factor. As a result, this design is confined to situations where the experimenter is not interested in predicting response at extremes (corners of the cube). A less

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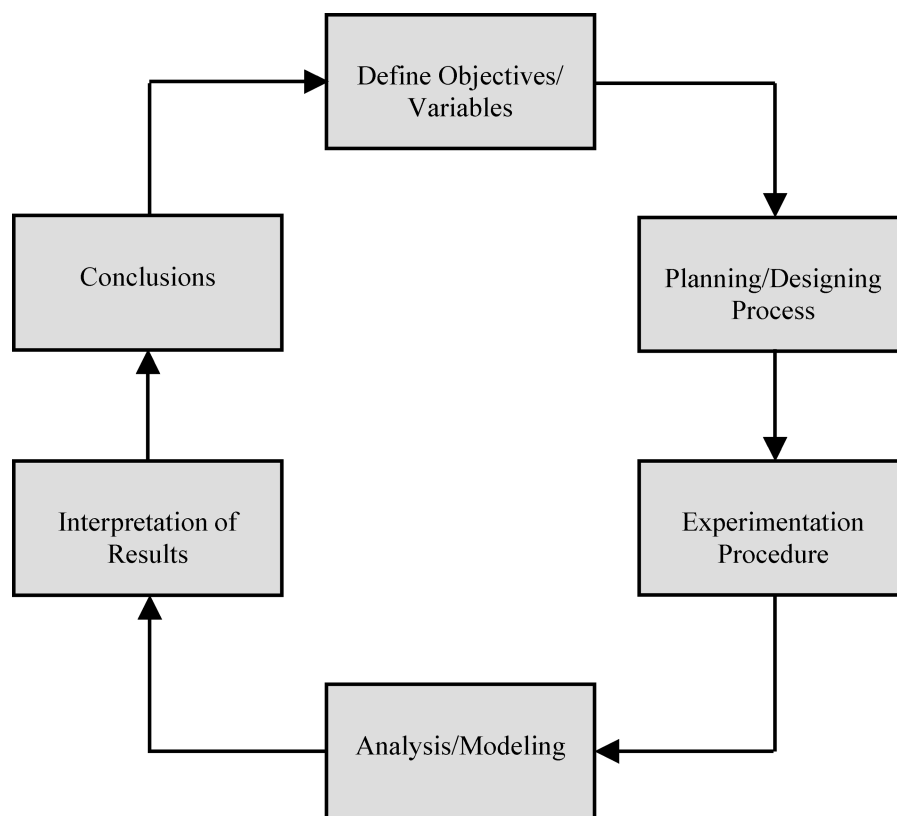


FIG. 1. Criteria for early experimental planning [modified from Hanrahan et al. (16) with permission from Elsevier].

common, but effective method is the Doehlert design. Like the Box Behnken design, Doehlert designs require lower numbers of experiments than the central composite design. Another advantage of the Doehlert design over the central composite approach is its higher efficiency value, ultimately determined by dividing the coefficient number of the quadratic equation by the number of experiments required for the design.

This paper critically examines the role that factorial and response surface methodology has played in modern (2000–Present) experimental design and optimization applications. Section one presents the theory and principles of factorial designs and response surface optimization methodology. This is followed by a detailed survey of recent applications including biological, environmental and pharmaceutical analysis, food technology and industrial-related processes. The last section summarizes the overall importance of experimental design and optimization techniques in modern research and development.

## TECHNIQUES AND APPLICATIONS

Table 1 lists examples of factorial and response surface methodology applications in modern experimental design and optimization. This list has been selected to provide a representative coverage of such techniques in biological, environmental and pharmaceutical analysis, food technology and industrial-

related processes. More in depth discussions in selected studies in each area are discussed below.

## Biological Applications

Novotná et al. (18) used full and fractional factorial designs in combination with artificial neural networks (ANN) in the optimization of high performance liquid chromatography (HPLC) separation of neuroprotective peptides. The ultimate goal of this research was to separate as many peptides as possible, before mass spectrometric analysis. This combination of fractional designs and ANN was novel in that it allowed the optimization of the separation conditions regardless of the limited structural and physico-chemical properties of analyzed peptides. The fractional factorial design proved successful in simple matrix separations and allowed ample input data for ANN approximation. After optimization, fractions with peptides were collected and analyzed using off-line matrix assisted laser desorption ionization time of flight mass spectrometry (MALDI-TOF-MS).

Hanrahan et al. (19) used a full factorial design with simplex optimization in the design and development of a flow injection-capillary electrophoresis (FI-CE) analyzer. Here, the binding of D-Ala-D-Ala terminus peptides to the antibiotic vancomycin (Van) was examined. A  $2^3$  full factorial design was employed to assess the impact of flow rate, injection time and voltage on instrument response (absorbance). Factors that had the greatest

TABLE 1  
Selected factorial and response surface methodology applications in modern experimental design and optimization

| Area of concentration | Experimental design and optimization techniques                  | Application   | Reference |
|-----------------------|--|---|-----------|
| Biological            | Full factorial, fractional factorial, artificial neural networks | Optimization of high performance liquid chromatography separation of neuroprotective peptides.  | 18        |
| Biological            | Full factorial, simplex  | Development of flow injection-capillary electrophoresis (FI-CE) to examine the binding of D-Ala-D-Ala terminus peptides to the antibiotic vancomycin. | 19        |
| Biological            | Fractional factorial   | The selection of main variables for improving <i>Tetrahymena thermophila</i> growth and enzyme production.  | 20        |
| Biological            | Full factorial   | Determination of experimental factors influencing the silanization of glass substrates in DNA chip technology.  | 21        |
| Biological            | Full factorial   | Designing a gas sensor based on extracts of <i>Pleurotus ostreatus</i> mycelium mushroom.   | 22        |
| Biological            | Full factorial   | Glyphosate determination using by square wave voltammetry.  | 23        |
| Biological            | Full factorial   | Production of antifungal antibiotic by <i>Thermomonospora</i> sp MTCC 3340.   | 24        |
| Biological            | Mechanistic, fuzzy and neural network components                 | Fed-batch fermentation optimization.  | 25        |
| Biological            | Central composite  | Optimization of micellar electrokinetic chromatography separations.   | 26        |
| Biological            | Full factorial   | Optimization of a plasmid DNA purification process.   | 27        |
| Biological            | Central composite  | High-density cDNA-microarray protocol optimization.   | 28        |
| Biological            | Full factorial, central composite                                | Optimization of the production of poly( $\gamma$ -glutamic acid) by <i>Bacillus licheniformis</i> CCRC 12826.   | 29        |
| Biological            | Central composite design   | Optimization of medium constituents for griseofulvin production.  | 30        |
| Biological            | Central composite design   | $\text{Fe}^{2+}$ and $\text{S}_2\text{O}_3^{2-}$ determination by an optimized amperometric bacterial sensor.   | 31        |
| Environmental         | Full factorial, central composite                                | Optimization of a fluorimetric method for the speciation of Cr(VI)/Cr(III).   | 32        |
| Environmental         | Full factorial, simplex  | Analysis of mercury species by commercially available instrumentation.  | 33        |
| Environmental         | Full factorial   | Determination of contaminants in Duero River by solid-phase microextraction.  | 34        |
| Environmental         | Fractional factorial, central composite                          | Optimization of a sequential extraction method for metal partitioning in soils and sediments.   | 35        |
| Environmental         | Fractional factorial   | Investigation of the influence of heavy metals and anions on cement hydration.  | 36        |
| Environmental         | Full factorial   | Voltammetric determination of lead in water samples.  | 37        |
| Environmental         | Full factorial   | Optimization of solid phase microextraction (SPME) conditions for butylated hydroxytoluene analysis.  | 38        |
| Environmental         | Full factorial, central composite                                | Flow-injection-hydride generation procedure for selenium determination.   | 39        |
| Environmental         | Full factorial, Doehlert   | On-line preconcentration and determination of zinc by flow injection inductively coupled plasma atomic emission spectrometry.                         | 40        |
| Environmental         | Full factorial   | Electrochemical treatment of methyl parathion.  | 41        |

(Continued on next page)

TABLE 1

Selected factorial and response surface methodology applications in modern experimental design and optimization (*Continued*)

| Area of concentration     | Experimental design and optimization techniques | Application   | Reference |
|---------------------------|---|---|-----------|
| Environmental             | Full factorial, Doehlert                        | Determination of copper using thermospray flame furnace atomic absorption spectrometry coupled to flow injection. | 42        |
| Environmental             | Full factorial                                  | The analysis of zinc, sodium, calcium and magnesium in water samples by capillary electrophoresis.                | 43        |
| Environmental             | Box–Behnken                                     | Determination of metallothionein by square wave cathodic stripping voltammetry.                                   | 44        |
| Environmental             | Full factorial, central composite               | Determination of chromium with ammonium pyrrolidine dithiocarbamate using adsorptive stripping voltammetry.       | 45        |
| Environmental             | Fractional factorial                            | Urinary benzene determination by SPME/GC-MS.  | 46        |
| Pharmaceutical            | Fractional factorial, central composite         | Screening of diuretics and ACE inhibitors using capillary zone electrophoresis.                                   | 47        |
| Pharmaceutical            | Fractional factorial, central composite         | Spectrofluorimetric determination of Losartan and Valsartan in human urine.                                       | 48        |
| Pharmaceutical            | Fractional factorial                            | Robustness testing of a flow-through dissolution method for atovaquone.   | 49        |
| Pharmaceutical            | Plackett–Burman, Doehlert                       | Capillary electrophoretic enantioresolution of salbutamol.  | 50        |
| Pharmaceutical            | Fractional factorial, central composite         | Development of spray-dried acetaminophen microparticles.  | 51        |
| Pharmaceutical            | Full factorial                                  | Determination of optimal amounts of water in hard gelatin capsules.   | 52        |
| Pharmaceutical            | Full factorial                                  | Production of diacetylmorphine/cafeine sachets.   | 53        |
| Pharmaceutical            | Full factorial                                  | Increasing bioavailability of silymarin using a buccal liposomal delivery system.                                 | 54        |
| Pharmaceutical            | Full factorial                                  | Development of an in vitro method for prediction of human drug absorption.  | 55        |
| Pharmaceutical            | Plackett–Burman, central composite              | Investigation of pharmaceutical high-performance liquid chromatography assay bias.                                | 56        |
| Pharmaceutical            | Full factorial                                  | Formation variables influencing the drug release rate from matrix tablets.  | 57        |
| Pharmaceutical            | Full factorial                                  | Design of experimental methods in modern pharmaceutical processes.  | 58        |
| Pharmaceutical            | Full factorial, fractional factorial            | Optimization of process parameters in small-scale fluidized bed granulation.                                      | 59        |
| Food/Industrial Processes | Full factorial, fractional factorial            | Experimental design methodology applied to optimize an organic synthesis.   | 60        |
| Food/Industrial Processes | Full factorial                                  | Analysis of the performance of a Proton Exchange Membrane Fuel Cell (PEMFC) stack.                                | 61        |
| Food/Industrial Processes | Full factorial                                  | Analysis of absorption-dehumidification processes.  | 62        |
| Food/Industrial Processes | Fractional factorial                            | Investigation of the properties of Ti/IrO <sub>2</sub> -Nb <sub>2</sub> O <sub>5</sub> electrodes.                | 63        |
| Food/Industrial Processes | Central composite, Plackett–Burman              | Optimization of a flow injection system for the determination of hydroquinone in cosmetics.                       | 65        |

*(Continued on next page)*

TABLE 1

Selected factorial and response surface methodology applications in modern experimental design and optimization (*Continued*)

| Area of concentration     | Experimental design and optimization techniques | Application  | Reference |
|---------------------------|---|--|-----------|
| Food/Industrial Processes | Doehlert  | Optimization of digestion procedures for the determination of Mn, Zn and Fe in food samples by FAAS. | 66        |
| Food/Industrial Processes | Central composite                               | Treatment of plasticized PVC to reduce plasticizer/solvent migration.                                | 67        |
| Food/Industrial Processes | Box–Behnken                                     | Determination of niacin in fresh and dry cured pork products by ion chromatography.                  | 68        |
| Food/Industrial Processes | Central composite                               | Fast GC analysis of major volatile compounds in distilled alcoholic beverages.                       | 69        |
| Food/Industrial Processes | Full factorial                                  | Optimization of coal flotation.  | 70        |
| Food/Industrial Processes | Full factorial                                  | Optimization of mechanical properties of polymer concrete and mix design.                            | 71        |
| Food/Industrial Processes | Central composite                               | Characterization of maize products based on their chromatographic profile.                           | 72        |
| Food/Industrial Processes | Central composite                               | Liquefaction of pine barks.  | 73        |

influence were flow rate and injection time. The simplex optimization method was used to further confirm the influencing factors and their optimum values. Optimization studies concluded the following optimum conditions: flow rate = 0.0625 mL min<sup>-1</sup>; injection time = 5 sec; voltage = 10,000 kV.

Figure 2 shows a representative series of electropherograms from the FI-CE analyzer for three injections of Van and NAD as standard in a capillary filled with increased concentrations of *N*-acetyl-D-Ala-D-Ala in the running buffer. The peaks for Van and the standard are baseline resolved and can easily be differentiated from each other at all concentrations and all injections. This paper details, for the first time, the use of FI-CE for the estimation of binding constants between a receptor and ligand utilizing affinity capillary electrophoresis (ACE). Moreover, it reports on the first known referenced application of experimental design methodology for FI-CE development.

### Environmental Applications

Massumi et al. (32) developed a fluorometric technique using Rhodamine-6G in the presence of H<sub>2</sub>SO<sub>4</sub> to study Cr(VI)/Cr(III) speciation in wastewater using chemometric experimental design and optimization methodology. A full factorial design was first implemented to evaluate the following factors: concentration of Rhodamine-6G (R), concentration of H<sub>2</sub>SO<sub>4</sub> (A), time of reaction (t) and temperature of reaction (T), on experimental response (intensity of fluorescence). Measurements were carried out for both low and high values and each effect was estimated by the Yates method. A central composite design was subsequently performed and coefficients calculated using the regression method.

The response surface plots generated are shown in Figures 3a and 3b, respectively. From the response surface, the optimum conditions were found to be at T = 109.15°C, t = 3 minutes, A = 0.415 M and R = 0.494 M. Cr(VI) and Cr(III) were measured in wastewater samples using the optimized technique. Overall, linear calibrations in the range of 8–50 ng mL<sup>-1</sup> Cr(VI) with a detection limit of 0.51 ng mL<sup>-1</sup> were shown at the optimum conditions from the central composite design results.

Prado et al. (46) used experimental design and response surface methodology in estimating the influence of experimental variables for biomonitoring benzene in exposed individuals. In this study, solid phase microextraction (SPME) and gas chromatography (GC) analysis was utilized in urinary benzene determination. In the screening portion, a half-fractional factorial design (five factors at two levels) was applied to ascertain the individual effects of sample temperature, incubation time, extraction time, sample volume and ionic strength on SPME extraction. The estimated effects of the factors and their interactions were calculated, with sample temperature and volume, incubation and extraction times significant. The study also showed significant interactions between these factors.

The most significant factors from the half-fractional factorial design were used to generate a response surface at three levels with an ANOVA performed to assess the significance of the model. Sample temperature and volume and their interactions had the largest effect on the response. Figure 4 shows the benzene area peak as a function of sample volume and sample temperature. As shown, increasing the sample temperature decreases the amount extracted. Low temperature had an obvious positive effect on the benzene area peak. The model predicted the following optimum factor conditions: 2.5 mL sample volume,

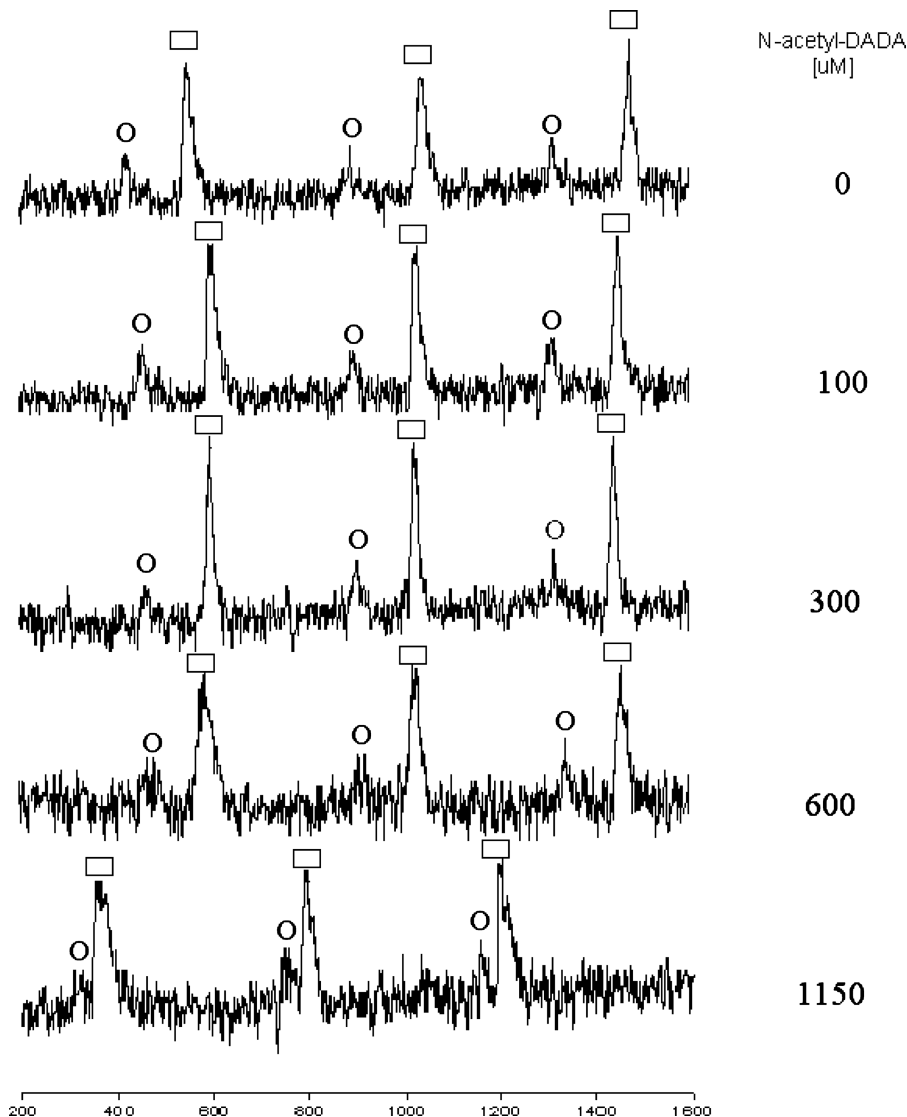


FIG. 2. A representative series of electropherograms of vancomycin (open circles) in 192 mM glycine–25 mM Tris buffer (pH 8.3) containing increasing concentrations of *N*-acetyl-D-Ala-D-Ala using the FI-CE instrument detailed herein. The total analysis time was 27 min at 10 kV (current 200  $\mu$ A) using a 60-cm (inlet to outlet), 74- $\mu$ m I.D. open, uncoated quartz capillary; flow rate = 0.0625 mL min<sup>-1</sup>, 5-second injection time with the detector was set at 220 nm. NAD (open rectangles) was used as an internal standard [from Hanrahan et al. (19) with permission from ISC Technical Publications].

15°C temperature, 1 minute extraction time and 1 minute incubation time. Overall, the optimized method showed good calibration linearity (typical  $r^2 = 0.999$ ) and sample reproducibility (typically <10% R.S.D.) with a detection limit of 0.043 ng mL<sup>-1</sup>.

### Pharmaceutical Applications

A novel spectrofluorimetric method for the determination of Losartan and Valsartan in human urine was developed with the aid of experimental design methodologies by Gagigal et al. (48). In this study, the influence of pH, buffer concentration, percentage of acetonitrile, temperature and slit width on the intrinsic

fluorescence needs of Losartan and Valsartan were examined to obtain maximum sensitivity for their determination. A cleanup procedure employing solid-phase extraction using C8 cartridges was used for all urine samples.

Fractional factorial ( $2^{6-3}$ ) results showed that the relative fluorescent response for both Losartan and Valsartan were negatively affected by increase pH values and positively affected by the increase of emission slit width. For Valsartan only, an increase in buffer concentration had a negative effect on the relative fluorescence response. In addition, the Valsartan relative fluorescence signal decreased as the temperature values increased.

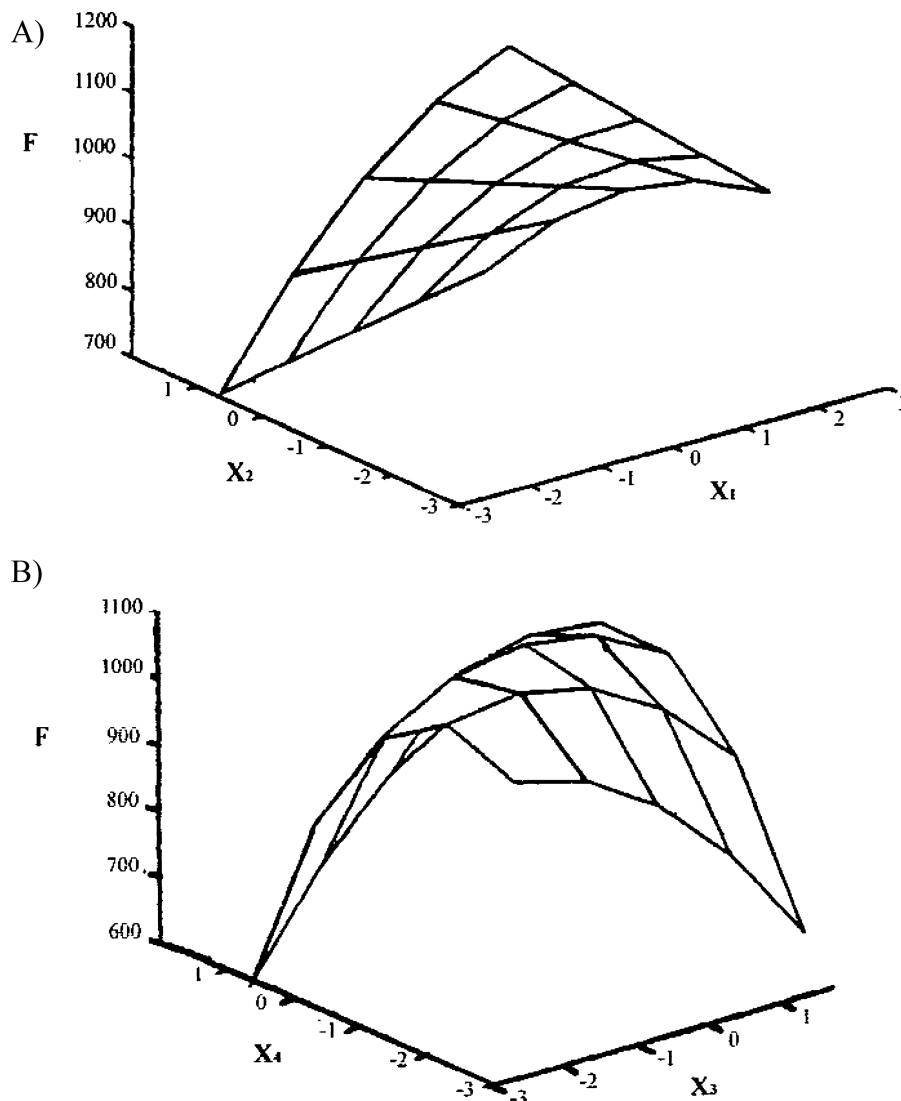


FIG. 3. (a) Plot of response function ( $\Delta F$ ) vs. temperature ( $X_1$ ) and time ( $X_2$ ) and (b) plot of response function ( $\Delta F$ ) vs. sulfuric acid concentration ( $X_3$ ) and R-6G concentration ( $X_4$ ) [from Massumi et al. (32) with permission from Elsevier].

Central composite designs employing two-level factorial designs plus star orthogonal composite designs involving 14 and 24 runs (plus three central points) were built based on the results of the fractional factorial designs for both Losartan and Valsartan, respectively. For Losartan, the relative fluorescence intensity decreased as the temperature increased. The emission slit width had a significant positive effect on the fluorescent response. In addition, interaction parameters for pH and temperature and pH and emission slit width were pronounced. For Valsartan, emission slit width had a significant positive effect over fluorescence response, the quadratic term and interactions with pH and temperature. However, buffer concentration had no significant interaction. Thus, it was determined that pH, temperature and emission slit width were the most important factors in both Losartan and Valsartan determination.

Overall, the use of experimental design methodology proved effective in optimizing the conditions for accurate (RE, 8%) and sensitive (LOQ c.a.  $0.5 \mu\text{g mL}^{-1}$ ) determination of Losartan and Valsartan in human urine. Total analysis time was less than 30 minutes, including the solid-phase extraction step and all results were corroborated by a complementary HPLC method.

Rambali et al. (59) used experimental design methodology to optimize the process parameters in fluidized bed granulation, the size enlargement step in the production of tablets in the pharmaceutical industries. Initially, a Plackett–Burman design was applied to screen the following parameters: spray rate ( $58.0\text{--}135.6 \text{ g min}^{-1}$ ), inlet air temperature ( $50\text{--}70^\circ\text{C}$ ), inlet flow rate ( $140\text{--}286 \text{ m}^3 \text{ h}^{-1}$ ), nozzle air pressure ( $1.5\text{--}2.5 \text{ bar}$ ), nozzle spray diameter ( $1.2\text{--}2.2 \text{ mm}$ ) and nozzle position ( $1.0\text{--}3.0$ ).



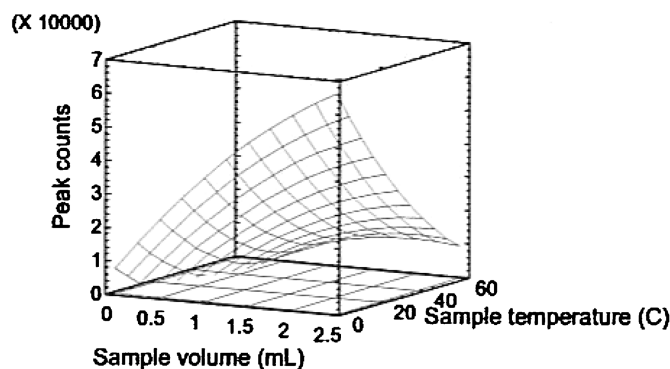


FIG. 4. Benzene area peak as a function of sample volume and sample temperature [from Prado et al. (46) with permission from Elsevier].

The significance of these factors on the percentage yield (w/w) of granules between 75 and 500  $\mu\text{m}$  and the geometric mean granule size ( $d_{50}$ ) were determined. The overall intent was to obtain granule yields of  $>90\%$  (between 75 and 500  $\mu\text{m}$ ) and a  $d_{50}$  between 300 and 500  $\mu\text{m}$ .

Results of the ANOVA for the Plackett–Burman design for the yield of granules showed that inlet air flow and spray rate had  $p$  values  $<0.05$ , indicating that they had a significant effect on the percentage yield and  $d_{50}$  value. A fractional factorial design ( $2^{5-2}$ ) was then employed to screen the remaining factors with the addition of nozzle aircap position and the spraying time interval. ANOVA results from the  $2^{5-2}$  fractional factorial confirmed the significance of spray rate and inlet air flow on both percentage yield and the  $d_{50}$  value. It also confirmed that nozzle air pressure had a significant effect ( $p < 0.05$ ) on the  $d_{50}$  value. Overall, optimum factor settings were: inlet air temperature =  $55^\circ\text{C}$ , spray rate =  $68.0 \text{ g min}^{-1}$ , nozzle air pressure = 2.5 bar, and inlet air flow rate =  $213 \text{ m}^3 \text{ h}^{-1}$ . As the target values were achieved, no further optimization techniques were employed.

### Food and Industrial Process Applications

Terezo and Pereira (63) used a  $2^{6-3}$  fractional factorial design in determining the effects of preparation factors on the electrochemical (anodic charge) and morphological properties of  $\text{Ti}/\text{IrO}_2\text{-Nb}_2\text{O}_5$  electrodes. The investigation of such electrochemical materials are important for the oxygen evolving reaction (OER), which occurs in such electrochemical processes as metal electrowinning, water electrolysis, cathodic protection and electroorganic reduction (64).

The preparation factors studied in this investigation included temperature, time and atmosphere of calcination, heating rate, molar ratio between citric acid/ethylene glycol (CA/EG) and citric acid/precursor salt (CA/PS). It was found that calcination temperature and composition of the precursor solution were the most significant factors affecting the anodic charge, most likely

due to the changes in coating morphology. More specifically, an average decrease of  $26.9 \text{ (mC cm}^{-2} \text{ mg}^{-1})$  was observed as the calcination temperature was changed from  $400$  to  $500^\circ\text{C}$ . The average anodic charge also decreased as the molar ratios between CA/EG (1:6 to 1:12) and CA/PS (6:1 to 12:1) were changed. Subsequent studies concentrated on the effects of calcination and heating rate in more detail. Here, a maximum anodic charge was observed at  $300^\circ\text{C}$ , while the electrode calcinated at  $600^\circ\text{C}$ .

Santelli et al. (66) used response surface methodology in the optimization of a focused microwave digestion procedure for the determination of Mn, Zn and Fe in food samples using flame atomic absorption spectrometry (FAAS). Here a Doehlert design involving three factors [irradiation power, irradiation time and percentage of oxidant solution— $\text{HNO}_3/\text{H}_2\text{O}_2$  (% v/v  $\text{H}_2\text{O}_2$ )] was developed through 13 experiments (including duplicate analysis at the central point to estimate experimental variance). In this design, irradiation time ( $T$ ) was studied at seven levels (2, 4, 6, 8, 10, 12, 14 minutes), irradiation power ( $P$ ) at five levels (60, 120, 180, 240, 300 W) and %  $\text{H}_2\text{O}_2$  at three levels (10, 30, 50).

In this study, the recovery (yield of focused microwave treatment) was calculated and used as the experimental response obtained from the analysis of commercially available food samples. The optimized values obtained from the Doehlert design using the focused microwave treatment were:  $T = 12$  minutes;  $P = 240\text{W}$  and %  $\text{H}_2\text{O}_2 = 42\%$ . The accuracy of the optimized method was evaluated by the analysis of certified reference materials and comparison with a closed vessel microwave oven dissolution method. Here, the application of a  $t$ -test showed no significant differences between the two methods and good recoveries were shown for all food samples using the focused microwave technique.

### CONCLUSIONS

Experimental design and optimization methodology are important in modern research and development efforts. In combination, these two strategies can help in optimizing experimental procedures in a reduced number of studies as well as providing essential information for appropriate decisions of the future of said procedures. This approach is opposite to the classical univariate approach. Univariate methods are time consuming in that the response is investigated for each factor while all other factors are held at a constant level. This approach is relatively simple and suitable for factors that are independent. However, univariate methods do not take interactive effects between factors into account. If the effects are additive in nature, then experimental designs are the optimum choice and require fewer measurements. In order to address the above concerns, and meet the demands of modern research, proper experimental design techniques considering all factors and their possible interactions must be performed.

# REFERENCES

1. R. M. Myers, *Response Surface Methodology* (Allyn and Bacon, Boston, 1971).
2. G. E. P. Box, W. G. Hunter, and J. S. Hunter, *Statistics for Experimenters: An Introduction to Design, Data Analysis and Model Building* (Wiley, New York, 1987).
3. S. N. Deming and S. L. Morgan, *Experimental Design: A Chemometric Approach* (Elsevier, Amsterdam, 1993).
4. S. D. Brown, and R. S. Bear, Chemometric techniques in electrochemistry: a critical review. *Critical Reviews in Analytical Chemistry* 24 (1993): 99–131.
5. J. Goupy, *Methods for Experimental Design. Principles and Applications for Physicists and Chemists* (Elsevier, Amsterdam, 1993).
6. R. Sundberg, Interpretation of unreplicated two-level factorial experiments, by examples. *Chemometrics and Intelligent Laboratory Systems* 24 (1994):1–17.
7. P. W. Araujo, and R. G. Brereton, Experimental design. I. Screening. *Trends in Analytical Chemistry* 15 (1996):26–31.
8. T. Lundstedt, E. Seifert, L. Abramo, B. Thelin, A. Nyström, J. Pettersen, and R. Bergman, Experimental design and optimization. *Chemometrics and Intelligent Laboratory Systems* 42 (1998):3–40.
9. C. H. Lockmüller, and C. E. Reese, Introduction to factor analysis. *Critical Reviews in Analytical Chemistry* 28 (1998):21–49.
10. D. L. Massart, B. G. M. Vandeginste, L. M. C. Buydens, S. de Jong, P. J. Lewi, and J. Smeyers-Verbeke, *Handbook of Chemometrics and Qualimetrics* (Elsevier, Amsterdam, 1998).
11. M. Otto, *Chemometrics: Statistics and Computer Applications in Analytical Chemistry* (Wiley-VCH, Chichester, 1999).
12. A. Dean and D. Voss, *Design and Analysis of Experiments* (Springer, New York, 1999).
13. D. R. Cox, and N. Reid, *Theory of the Design of Experiments* (Chapman and Hall, New York, 2000).
14. J. N. Miller and J. C. Miller, *Statistics and Chemometrics for Analytical Chemistry*, 4th Edition (Prentice-Hall, Englewood Cliffs, NJ, 2000).
15. M. F. W. Festing, Principles: The need for better experimental design. *Trends in Pharmacological Sciences* 24 (2003):341–345.
16. P. W. Araujo and R. G. Brereton, Experimental design. II. Optimization. *Trends in Analytical Chemistry* 15 (1996):63–70.
17. G. Hanrahan, J. Zhu, S. Gibani, and D. G. Patil, Chemometrics: experimental design, in *Encyclopedia of Analytical Science*, eds. P. J. Worsfold, A. Townshend, and C. F. Poole (Elsevier, Oxford, 2005), Vol. 2, 8–13.
18. K. Novatná, J. Havliš, and J. Havel, Optimisation of high performance liquid chromatography separation of neuroprotective peptides. Fractional experimental designs combined with artificial neural networks. *Journal of Chromatography A* 1096 (2005):50–57.
19. G. Hanrahan, F. Tse, F. T. Dahdouh, K. Clarke, and F. A. Gomez, The design and development of a flow injection-capillary electrophoresis (FI-CE) analyzer employing fiber optic detection. *Journal of Capillary Electrophoresis and Microchip Technology*, in press.
20. J. De Coninck, B. Leclercq, J. M. Exbrayat, and F. Duyme, Factorial designs: an efficient approach to choosing the main factors influencing growth and hydrolase production by *Tetrahymena thermophila*. *Journal of Industrial Microbiological Technology* 31 (2004):204–208.
21. C. M. Halliwell and A. E. G. Cass, A factorial analysis of silanization of conditions for the immobilization of oligonucleotides on glass surfaces. *Analytical Chemistry* 73 (2001):2476–2483.
22. Y. E. Silina, T. A. Kuchmenko, Y. I. Korenman, O. M. Tsvileva, and V. E. Nikitina, Use of a complete factorial experiment for designing a gas sensor based on extracts of *Pleurotus ostreatus* mycelium mushroom. *Journal of Analytical Chemistry* 60 (2005):759–764.
23. R. F. Teofilo, E. L. Reis, C. Reis, G. A. da Silva, and L. Kubota, Experimental design employed to square wave voltammetry response optimization for glyphosate determination. *Journal of the Brazilian Chemical Society* 15 (2004):865–871.
24. M. Gupte and P. Kulkarni, A study of antifungal antibiotic production by *Thermomonospora* sp MTCC 3340 using full factorial design. *Journal of Chemical Technology and Biotechnology* 78 (2003):605–610.
25. R. Berkholz, D. Röhlig, and R. Guthke, Data and knowledge based experimental design for fermentation process optimization. *Enzyme and Microbial Technology* 27 (2000):784–788.
26. J. C. Ehlen, H. E. Albers, and E. D. Breyer, MEKC-LIF of  $\gamma$ -amino butyric acid in microdialysate: systematic optimization of the separation conditions by factorial analysis. *Journal of Neuroscience Methods* 147 (2005):36–47.
27. A. Eon-Duval, K. Gumbs, and C. Ellett, Precipitation of RNA impurities with high salt in a plasmid DNA purification process: use of experimental design to determine reaction conditions. *Biotechnology and Bioengineering* 83 (2003):544–553.
28. G. Wrobel, J. Schlingemann, L. Hummerich, H. Kramer, P. Lichter, and M. Hahn, Optimization of high-density cDNA-microarray protocols by design of experiments. *Nucleic Acids Research* 31 (2003):67–73.
29. I. L. Smith, Y. T. Van, and Y. N. Chang, Application of statistical experimental methods to optimize production of poly( $\gamma$ -glutamic acid) by *Bacillus licheniformis* CCRC 12826. *Enzyme and Microbial Technology* 31 (2002):213–220.
30. V. Venkata, T. Panda, and M. Chidambaram, Development of medium for griseofulvin production: part II. Optimization of medium constituents using central composite design. *Journal of Microbiology and Biotechnology* 12 (2002):360–366.
31. R. Zlatev, J.-P. Magnin, P. Ozil, and M. Stoytcheva, Bacterial sensors based on *Acidithiobacillus ferrooxidans* part 1.  $\text{Fe}^{2+}$  and  $\text{S}_2\text{O}_3^{2-}$  determination. *Biosensors and Bioelectronics* 21 (2006):1493–1500.
32. A. Massumi, N.M. Najafi, and H. Barzegari, Speciation of Cr(VI)/Cr(III) in environmental waters by fluorimetric method using central composite, full and fractional factorial design. *Microchemical Journal* 72 (2002):93–101.
33. J. Sanz, A. de Diego, J. C. Raposo, and J. M. Madariaga, Routine analysis of mercury species using commercially available instrumentation: chemometric optimization of the instrumental variables. *Analytica Chimica Acta* 486 (2003):55–267.
34. J. Salafranca, C. Domeno, C. Fernández, and C. Nerín, Experimental design applied to the determination of several contaminants in Duero River by solid-phase microextraction. *Analytical Chimica Acta* 477 (2003):257–267.
35. R. Santamaria-Fernández, A. Moreda-Piñeiro, and S. J. Hill, Optimization of a multielement sequential extraction method employing an experimental design approach for metal partitioning

- in soils and sediments. *Journal of Environmental Monitoring* 4 (2002):330–336.
36. A. Poletini, R. Pomi, and P. Sirini, Fractional factorial design to investigate the influence of heavy metals and anions on acid neutralization behavior of cement-based products. *Environmental Science and Technology*, 36 (2002):1584–1591.
  37. A. A. Ensafi, T. Khayamian, and M. Atabati, Differential pulse cathodic stripping adsorption voltammetric determination of trace amounts of lead using factorial design for optimization. *Talanta* 59 (2003):727–733.
  38. N. B. Tombesi, R. H. Freije, and F. Augusto, Factorial experimental design optimization of solid phase microextraction (SPME) conditions for analysis of butylated hydroxytoluene (BHT) in bottled water. *Journal of the Brazilian Chemical Society* 15 (2004):658–663.
  39. N. Etxebarria, R. Antolín, G. Borge, T. Posada, and J. C. Raposo, Optimization of flow-injection-hydride generation inductively coupled plasma spectrometric determination of selenium in electrolytic manganese. *Talanta* 65 (2005):1209–1214.
  40. M. Zougagh, P. C. Rudner, A.G. de Torres, and J.M.C. Pavón, Application of Doehlert matrix and factorial designs in the optimization of experimental variables associated with the on-line preconcentration and determination of zinc by flow injection inductively coupled plasma atomic emission spectrometry. *Journal of Analytical and Atomic Spectroscopy* 15 (2000):1589–1594.
  41. A. G. Vlyssides, D. G. Arapoglou, C. J. Israilides, E. M. P. Barampouti, and S. T. Mai, Electrochemical treatment of methyl parathion based on the implementation of a factorial design. *Journal of Applied Electrochemistry* 34 (2004):1265–1269.
  42. C. R. T. Tarley and E. C. Figueiredo, G. D. Matos, Thermospray flame furnace-AAS determination of copper after on-line sorbent preconcentration using a system optimized by experimental designs. *Analytical Sciences* 21 (2005):1337–1342.
  43. J. A. Jurado-González, M. D. Galindo-Riaño, and M. García-Vargas, Factorial designs applied to the development of a capillary electrophoresis method for the analysis of zinc, sodium, calcium and magnesium in water samples. *Talanta* 59 (2003):775–783.
  44. M. El Hourch, A. Dudoit, and J.-C. Amiard, An optimization procedure for determination of metallothionein by square wave cathodic stripping voltammetry: Application to marine worms. *Analytical and Bioanalytical Chemistry* 378 (2004):776–781.
  45. O. Domínguez, M. Asunción, and M. J. Arcos, Application of an optimization procedure in adsorptive stripping voltammetry for the determination of chromium with ammonium pyrrolidine dithiocarbamate. *Electroanalysis* 14 (2002):1083–1089.
  46. C. Prado, J. Garrido, and J. F. Periago, Urinary benzene determination by SPME/GC-MS. A study of variables by fractional factorial design and response surface methodology. *Journal of Chromatography B* 804 (2004):255–261.
  47. U. Akesolo, M. I. Maguregui, L. González, R. M. Jiménez, and R. M. Alonso, Experimental design optimization of a capillary zone electrophoresis method for the screening of several diuretics and ACE inhibitors. *Journal of Chromatographic Science* 42 (2004):74–79.
  48. E. Cagigal, L. González, R. M. Alonso, and R. M. Jiménez, Experimental design methodologies to optimize the spectrofluorimetric determination of Losartan and Valsartan in human urine. *Talanta* 54 (2001):1121–1133.
  49. M. S. Bloomfield and W. C. Butler, Robustness testing, using experimental design, of a flow-through dissolution method for a product where the actives have markedly differing solubility properties. *International Journal of Pharmaceutics* 206 (2000):55–61.
  50. R. Gotti, S. Furlanetto, V. Andrisano, V. Cavrini, and S. Pinzauti, Design of experiments for capillary electrophoretic enantioresolution of salbutamol using dermatan sulfate. *Journal of Chromatography A* 875 (2000):411–422.
  51. A. Billon, B. Bataille, G. Cassanas, and M. Jacob, Development of spray-dried acetaminophen microparticles using experimental designs. *International Journal of Pharmaceutics* 203 (2000):159–168.
  52. M. Kuentz and D. Röthlisberger, Determination of the optimal amount of water in liquid-fill masses of hard gelatin capsules by means of texture analysis and experimental designs. *International Journal of Pharmaceutics* 236 (2002):145–152.
  53. M. G. Klous, B. Nuijen, W. Van den Brink, J. M. Van Ree, and J. H. Beijnen, Process characterisation, optimisation and validation of production of diacetylmorphine/cafeine sachets: a design of experiments approach. *International Journal of Pharmaceutics* 285 (2004):65–75.
  54. M. S. El-Samaliy, N. N. Afifi, and E. A. Mahmoud, Increasing bioavailability of silymarin using buccal liposomal delivery system: Preparation and experimental design investigation. *International Journal of Pharmaceutics* 308 (2006):140–148.
  55. G. Corti, F. Maestrelli, M. Cirri, S. Furlanetto, and P. Mura, Development and evaluation of an in vitro method for prediction of human drug absorption I. Assessment of artificial membrane composition. *European Journal of Pharmaceutical Sciences* 27 (2006):346–353.
  56. J.-G. Chen, K. Glancy, X. Chen, and M. Alasandro, Investigation of pharmaceutical high-performance liquid chromatography assay bias using experimental design. *Journal of Chromatography A* 917 (2001):63–73.
  57. S. Furlanetto, M. Cirri, F. Maestrelli, G. Corti, and P. Mura, Study of formulation variables influencing the drug release rate from matrix tablets by experimental design. *European Journal of Pharmaceutics and Biopharmaceutics* 62 (2006):77–84.
  58. O. W. Gooding, Process optimization using combinatorial design principles: parallel synthesis and design of experiment methods. *Current Opinion in Chemical Biology* 8 (2004):297–304.
  59. B. Rambali, L. Baert, D. Thoné, and D. L. Massart, Using experimental design to optimize the process parameters in fluidized bed granulation. *Drug Development and Industrial Pharmacy* 27 (2001):47–55.
  60. J. Guervenou, P. Giamarchi, C. Coulouarn, M. Guerda, C. Le Lez, and T. Oboyet, Experimental design methodology and data analysis technique applied to optimize an organic synthesis. *Chemometrics and Intelligent Laboratory Systems* 63 (2002):81–89.
  61. M. F. Torchio, M. G. Santarelli, and A. Nicali, Experimental analysis of the CHP performance of a PEMFC stack by a 2<sup>4</sup> factorial design. *Journal of Power Sources* 149 (2005):33–43.
  62. M.-H. Lai, Y.-N. Chang, C.-M. Wang, H. Wu, and T.-W. Chung, Analysis of the absorption-dehumidification process variables using the experimental design methodology. *Separation Science and Technology* 38 (2003):2447–2464.
  63. A. J. Terezo and E. C. Pereira, Fractional factorial design applied to investigate properties of Ti/IrO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub> electrodes. *Electrochimica Acta* 45 (2000):4351–4358.

64. A. Nidola, *Electrodes of Conductive Metallic Oxides, Part B* (Elsevier, Amsterdam, 1981).
65. M. E. Rueda, L. A. Sarabia, A. Herrero, and M. C. Ortiz, Optimisation of a flow injection system with electrochemical detection using the desirability function. Application to the determination of hydroquinone in cosmetics. *Analytica Chimica Acta* 479 (2003):173–184.
66. R. E. Santelli, M. de Almeida Bezerra, O. D. de SantAna, R. J. Cassella, and S. L. C. Ferreira, Multivariate technique for optimization of digestion procedure by focused microwave system for determination of Mn, Zn and Fe in food samples using FAAS. *Talanta* 68 (2006):1083–1088.
67. J.-L. Fugit, J.-L. Taverdet, J.-V. Gauvrit, and P. Lanteri, Treatment of plasticized PVC to reduce plasticizer/solvent migration: optimization with an experimental design. *Polymer International* 52 (2003):670–675.
68. G. Sacconi, E. Tanzi, S. Mallozzi, and S. Cavalli, Determination of niacin in fresh and dry cured pork products by ion chromatography: experimental design approach for the optimization of nicotinic acid separation. *Food Chemistry* 92 (2005):373–379.
69. K. Mac Namara, R. Leardi, and A. Sabuneti, Fast GC analysis of major volatile compounds in distilled alcoholic beverages. Optimisation of injection and chromatographic conditions. *Analytica Chimica Acta* 542 (2005):260–267.
70. P. K. Naik, P. S. R. Reddy, and V. N. Misra, Optimization of coal flotation using statistical technique. *Fuel Processing Technology* 85 (2004):1473–1485.
71. M. Muthukumar and D. Mohan, Optimization of mechanical properties of polymer concrete and mix design recommendation based on design of experiments. *Journal of Applied Polymer Science* 94 (2004):1107–1116.
72. J. M. Rodriguez-Nogales, M. C. Garcia, and M. L. Marina, Monolithic supports for the characterization of commercial maize products based on their chromatographic profile. Application of experimental design and classification techniques. *Journal of Agricultural and Food Chemistry* 54 (2006):1173–1179.
73. K. Acikalin, F. Karaca, and E. Bolat, Central composite rotatable design for liquefaction of pine barks. *Fuel Processing Technology* 87 (2005):17–24.